

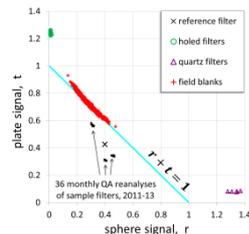
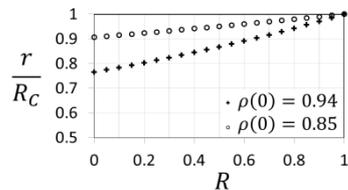
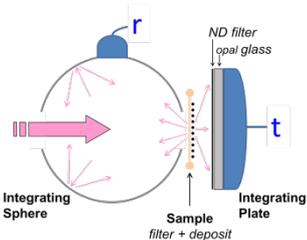
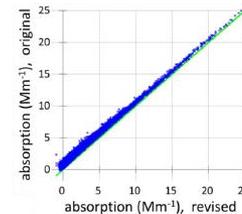
HIPS data are now reported with a consistent calibration (2003 – present)

DRAFT, 10/12/2015

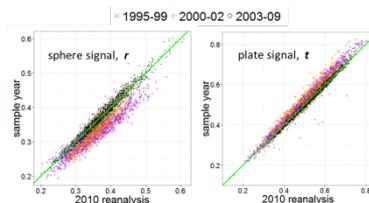
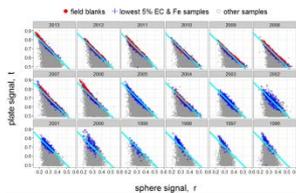
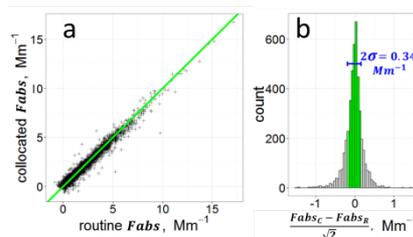
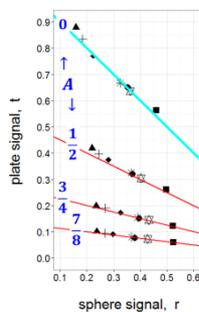
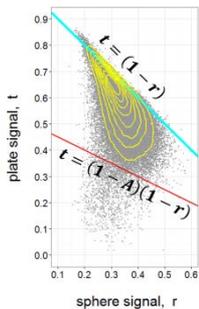
Light absorption by aerosol deposits on PTFE filters: a decade of backscatter-corrected transmittance measurements by the IMPROVE regional haze network

Warren H. White, Krystyna Trzepla, Nicole P. Hyslop
Crocker Nuclear Laboratory, University of California, Davis, CA

Posting type Informational
Subject Changed reporting of light absorption
Module/Species A/ Fabs
Sites Entire network
Period Data downloaded after 2014 for all sample years
Submitter W.H. White, whwhite@ucdavis.edu

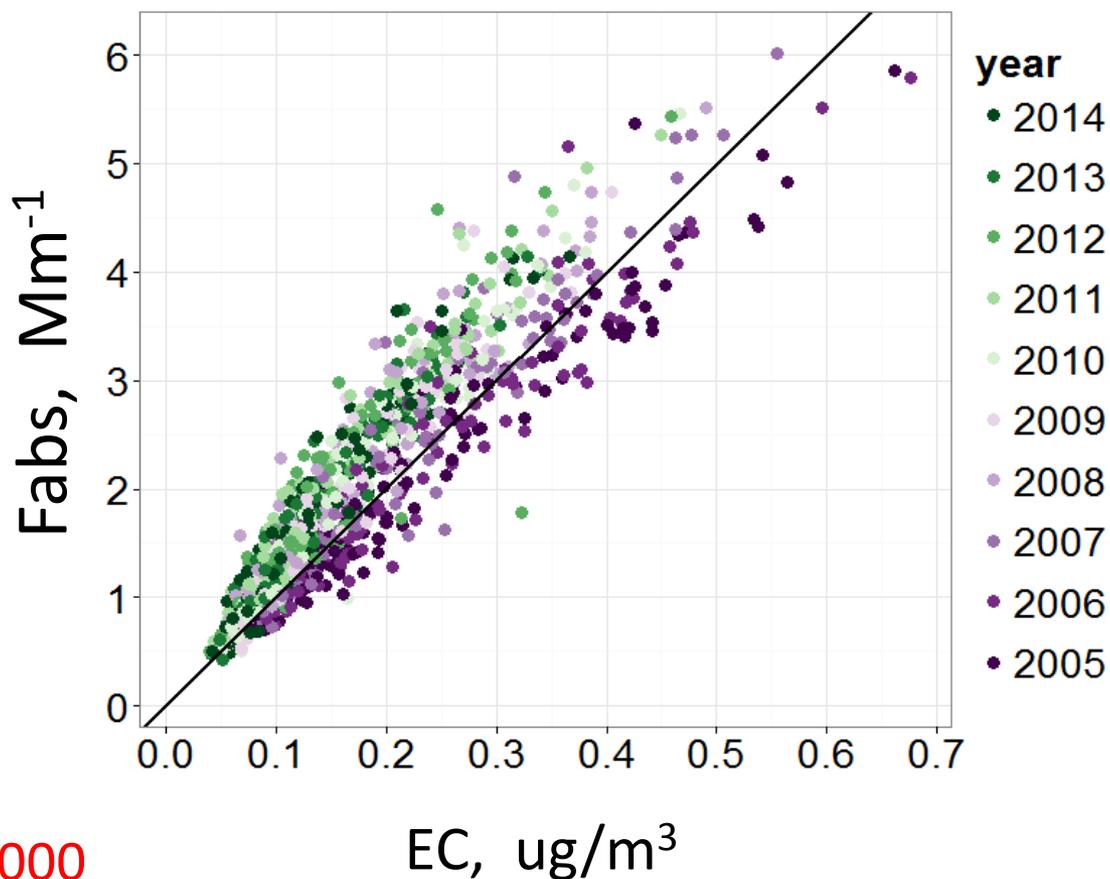


So ...
what do
they say?



Well, *Fabs* correlates well with *EC* (!!), as seen below with site-year means. Each point shows the arithmetic means from at least 100 observations during one calendar year at one of 116 IMPROVE sites selected for data homogeneity:

- ✓ unmasked A module
- ✓ non-urban location
- ✓ $-130 < \text{longitude} < -65$
- ✓ $|FM-RM| < 5 \text{ ug/m}^3$ *
- ✓ at least 1000 samples from 2005 – 2014



* excludes fewer than 3 in 1000

$$RM \equiv \frac{132}{96} \cdot SO_4 + \frac{80}{62} \cdot NO_3 + 1.65 \cdot Cl + EC + 1.7 \cdot OC + 20 \cdot Fe$$

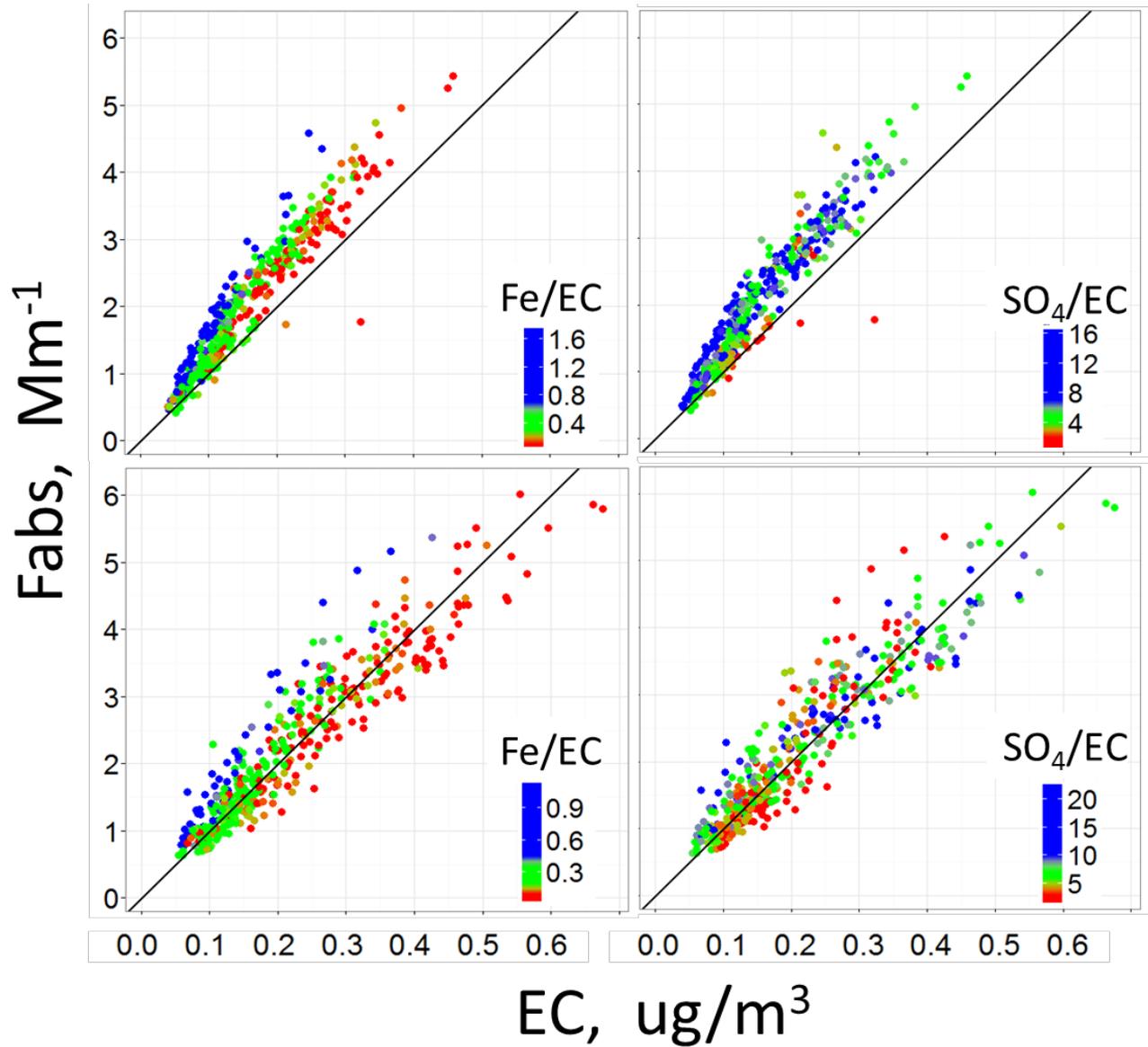
Question:

Where does the scatter come from?

Partial

Answer 1:

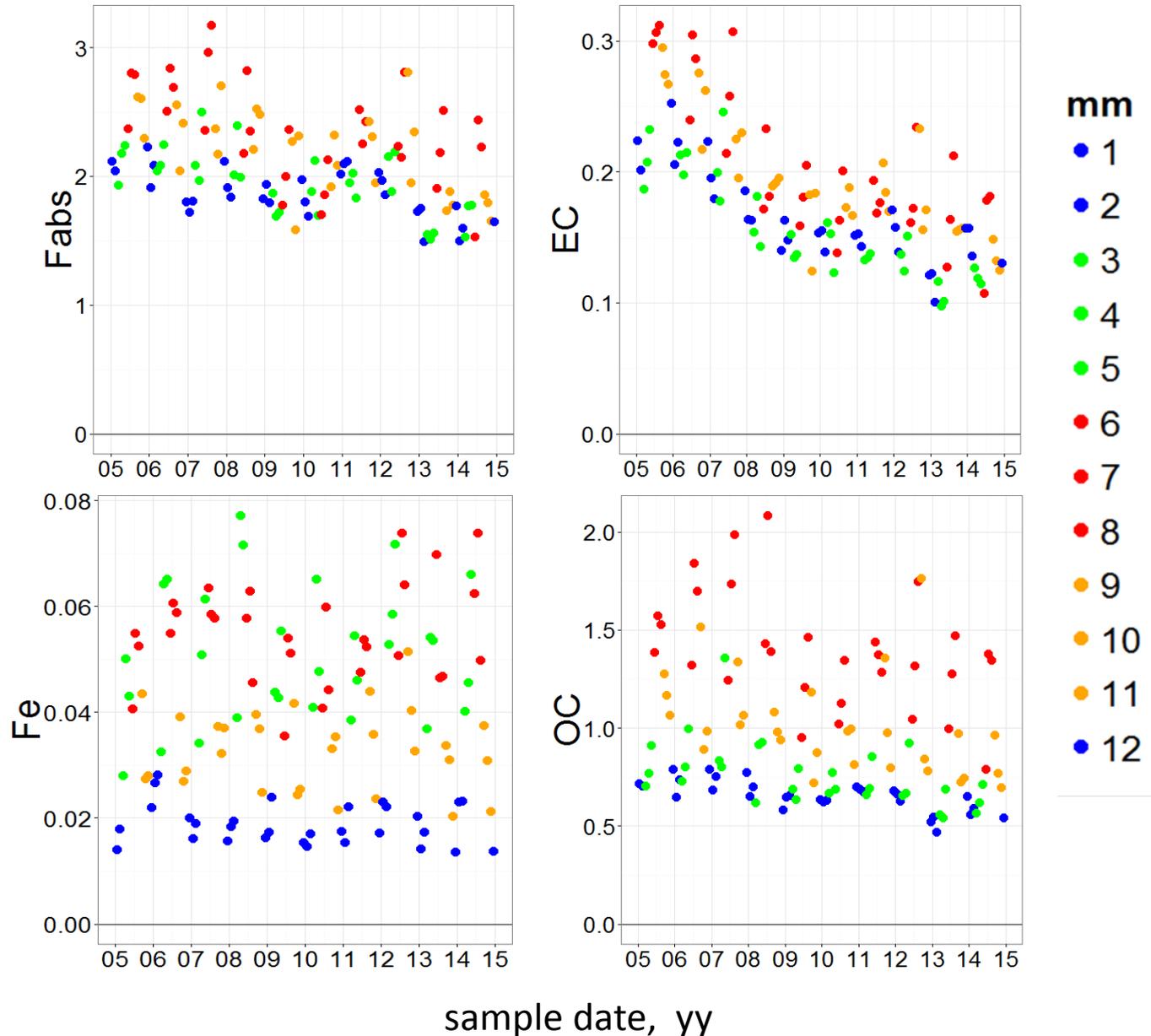
EC is not the sole absorbing species. Iron oxide minerals also absorb, for example. Sulfates don't.



month-mean 116-site concentrations, Mm^{-1} or $\mu\text{g}/\text{m}^3$

A1, continued:

Fabs has not declined as fast as EC has because other absorbers are not declining as fast as EC, ...



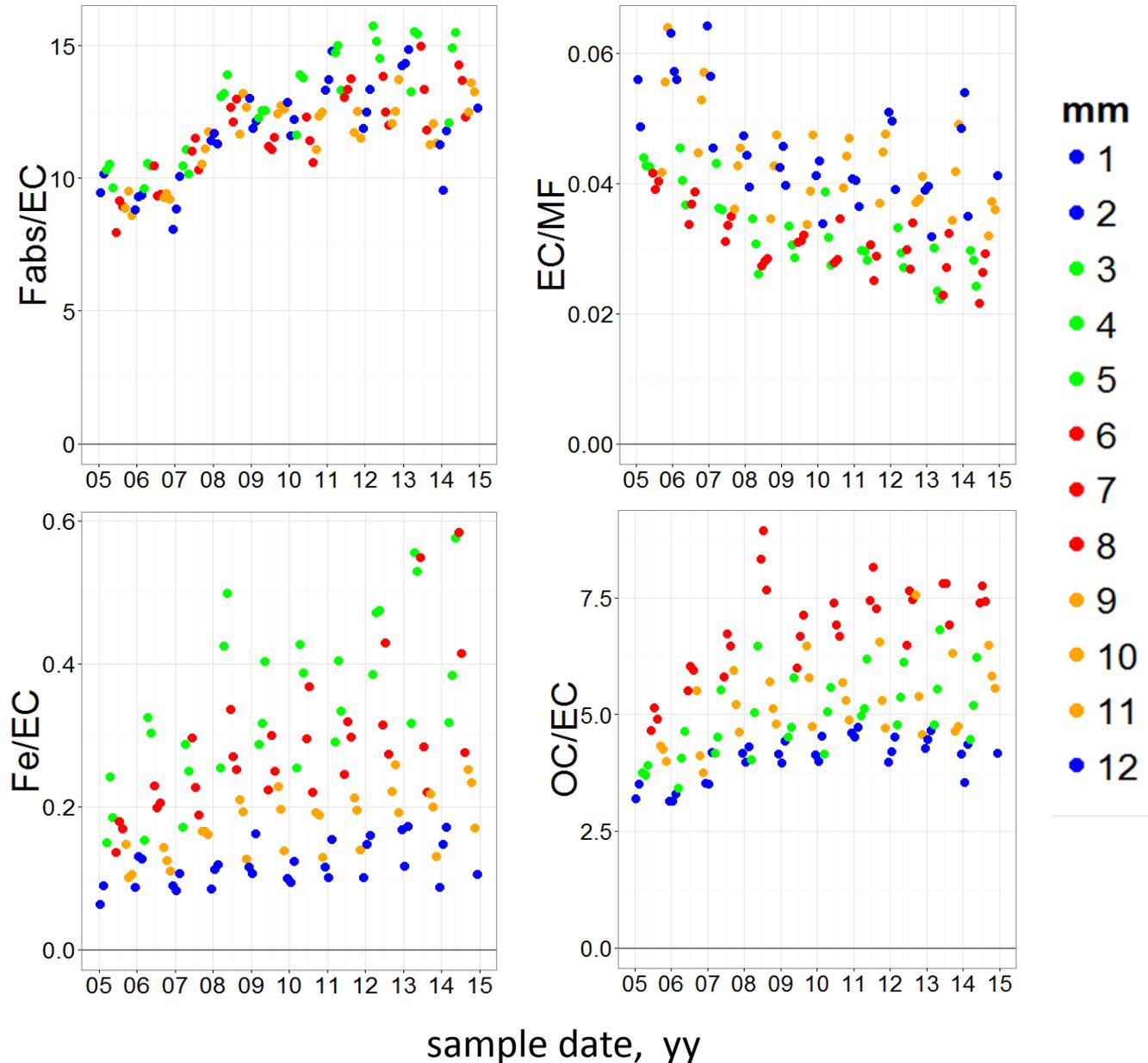
ratios of month-mean concentrations, m^2/h or g/g

A1, continued:

... so total absorption increasingly reflects dust and other species. *But this is only part of what is changing.*

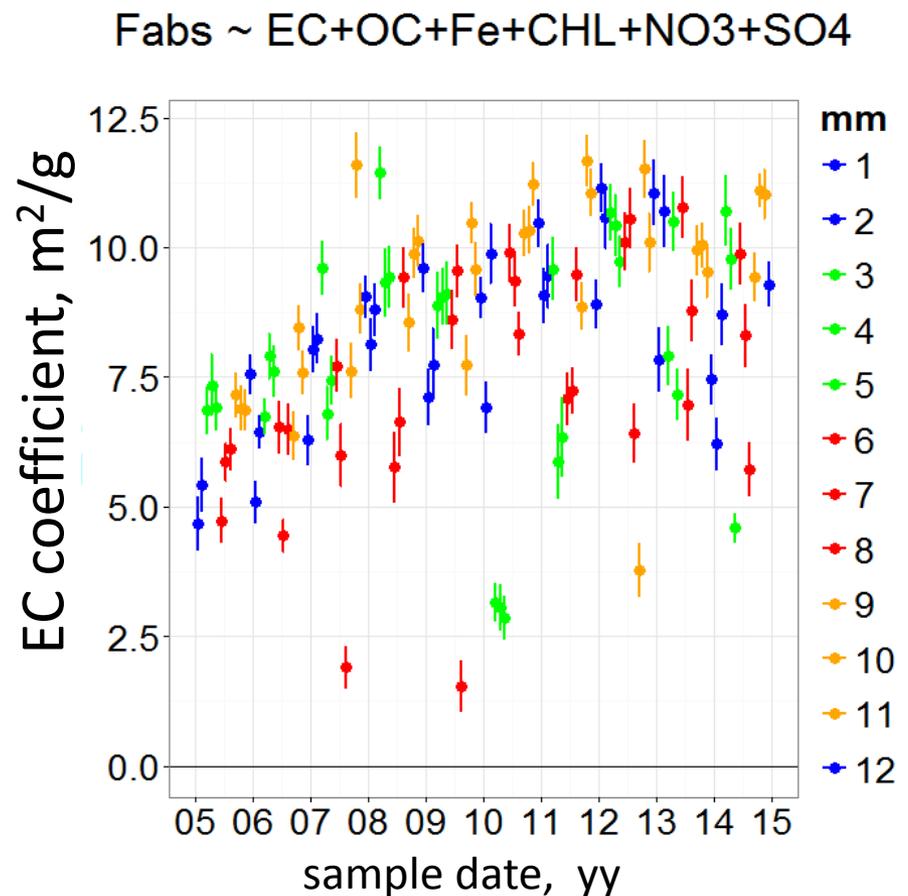
Partial answer 2:

LAC is itself complex and evolving.



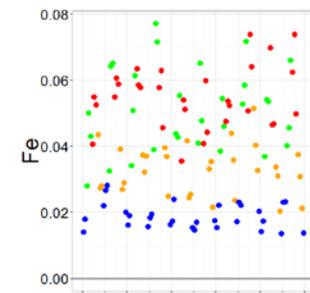
Multiple linear regression analysis is a convenient approach to assess the sensitivity of total absorption to individual chemical species. It provides only a statistical model, not a causal relationship, as the contributions of different species need not be additive (e.g., Fuller, Malm and Kreidenweis, 1999).

OLS (classical) regressions were performed a month at a time, in 120 separate calculations. Coefficients are plotted with 95% confidence intervals. The regression model is indicated above the plot, using conventions of the R language.



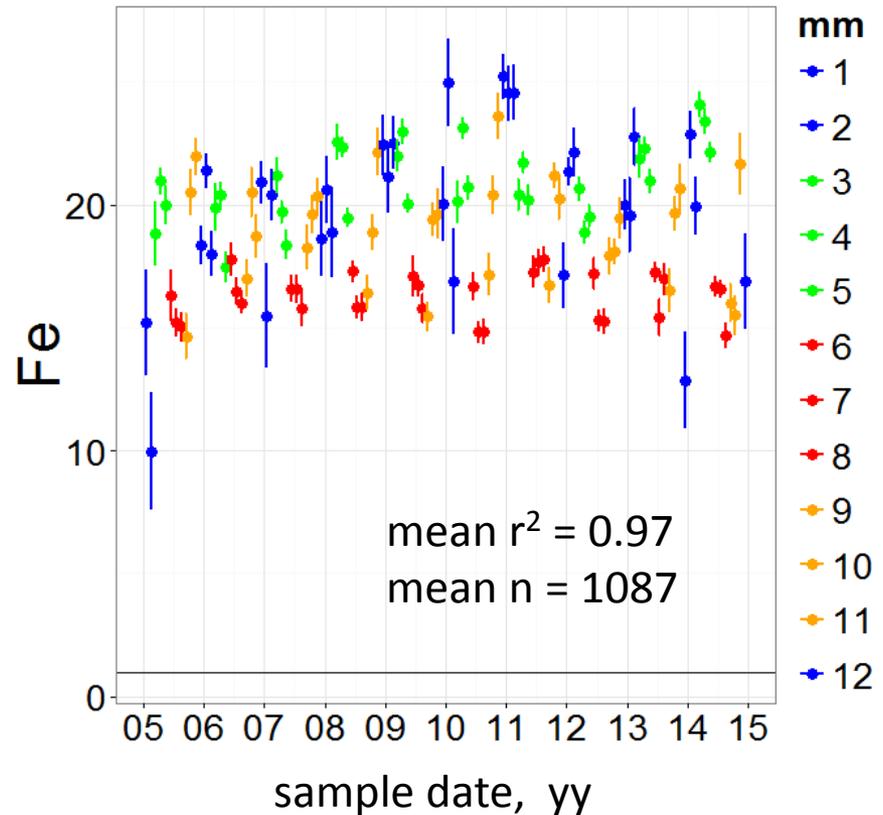
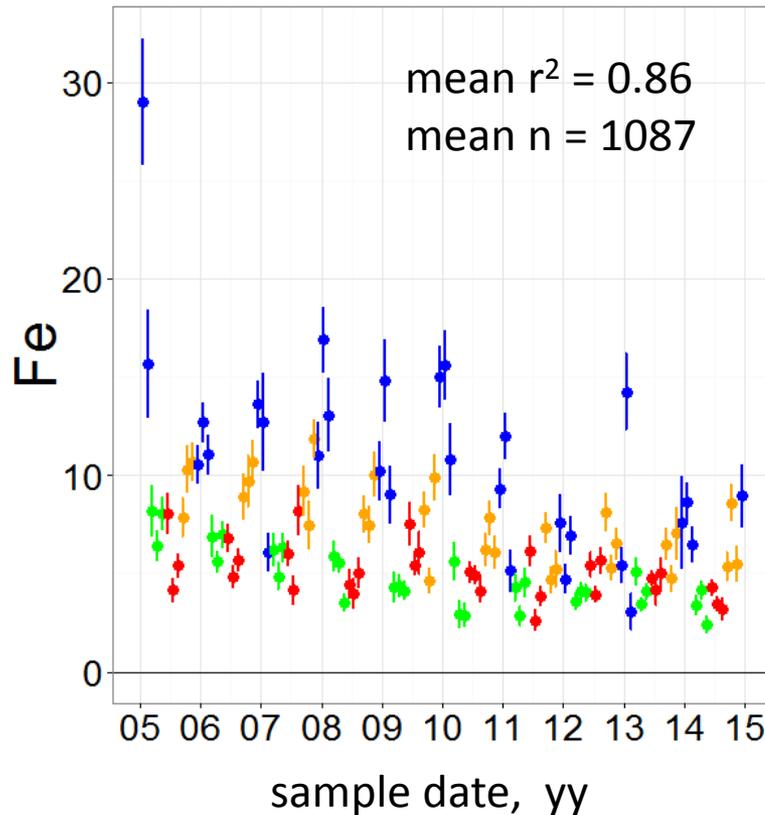
Before digging into LAC, we'll first "calibrate" the regression approach on other species.

Recall the seasonality of Fe concentrations →



Fabs ~ EC+OC+Fe+CHL+NO3+SO4

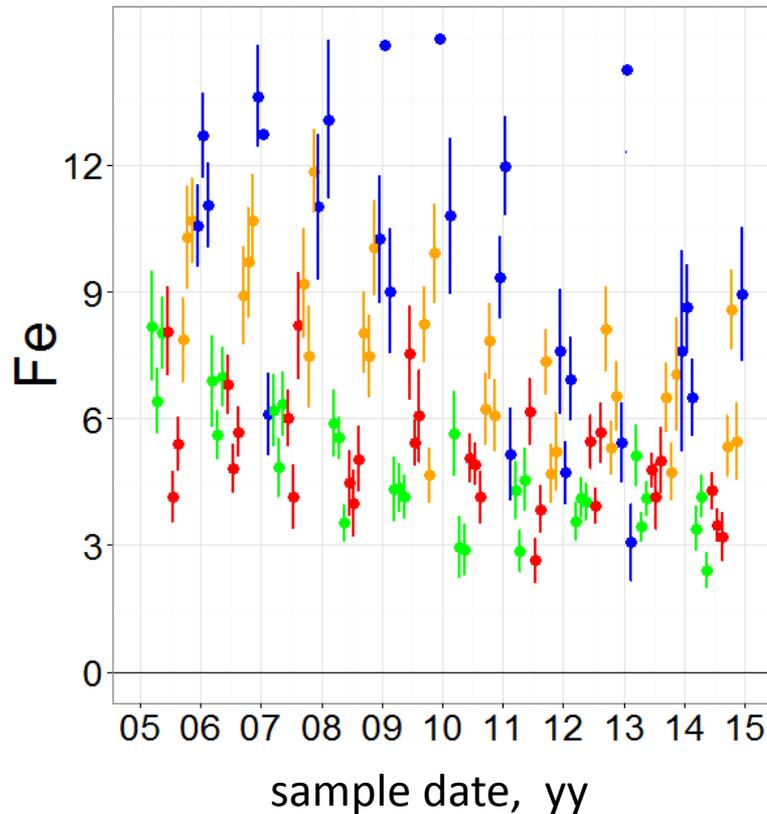
MF-EC ~ OC+Fe+CHL+NO3+SO4



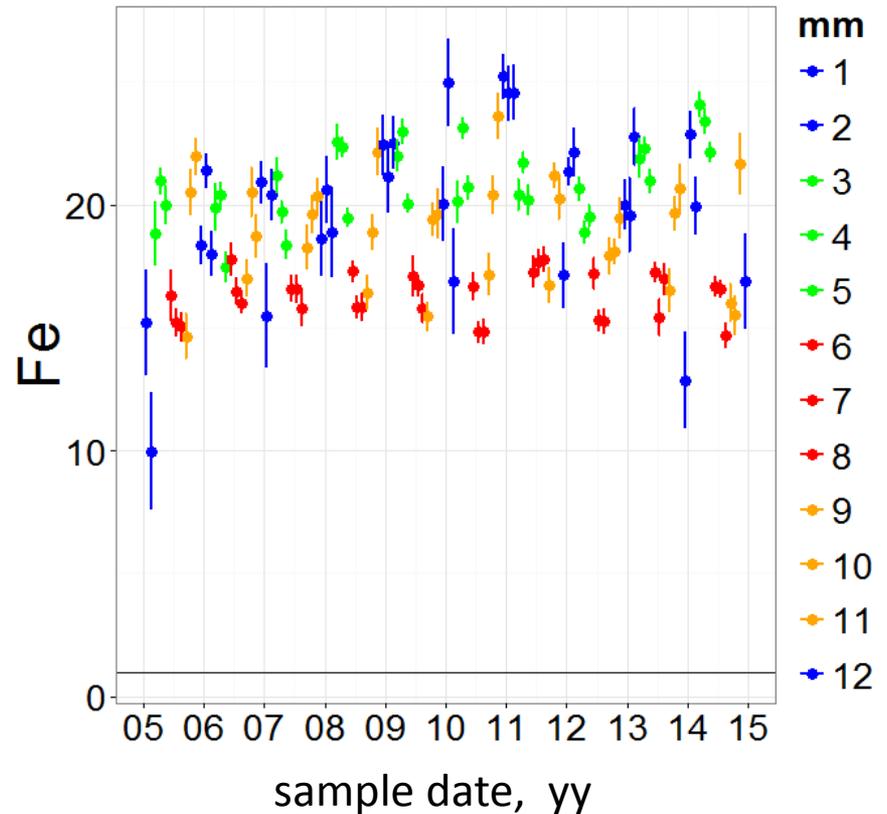
OLS coefficients from monthly regressions, m^2/g or g/g

The coefficients for the winter months are poorly determined, suggesting that the absorption plot be re-scaled for better visibility.

Fabs ~ EC+OC+Fe+CHL+NO3+SO4



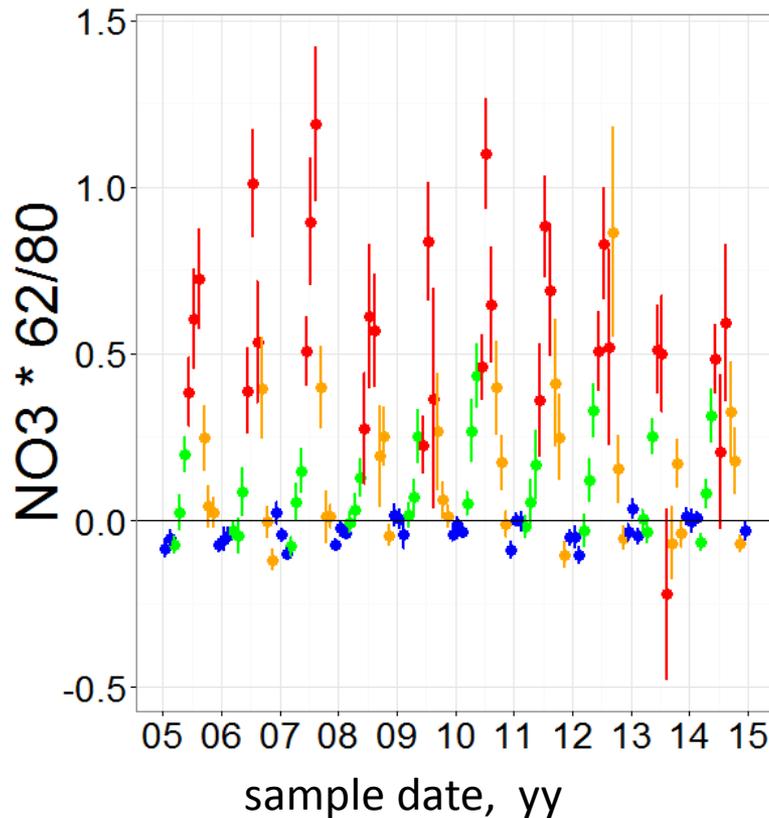
MF-EC ~ OC+Fe+CHL+NO3+SO4



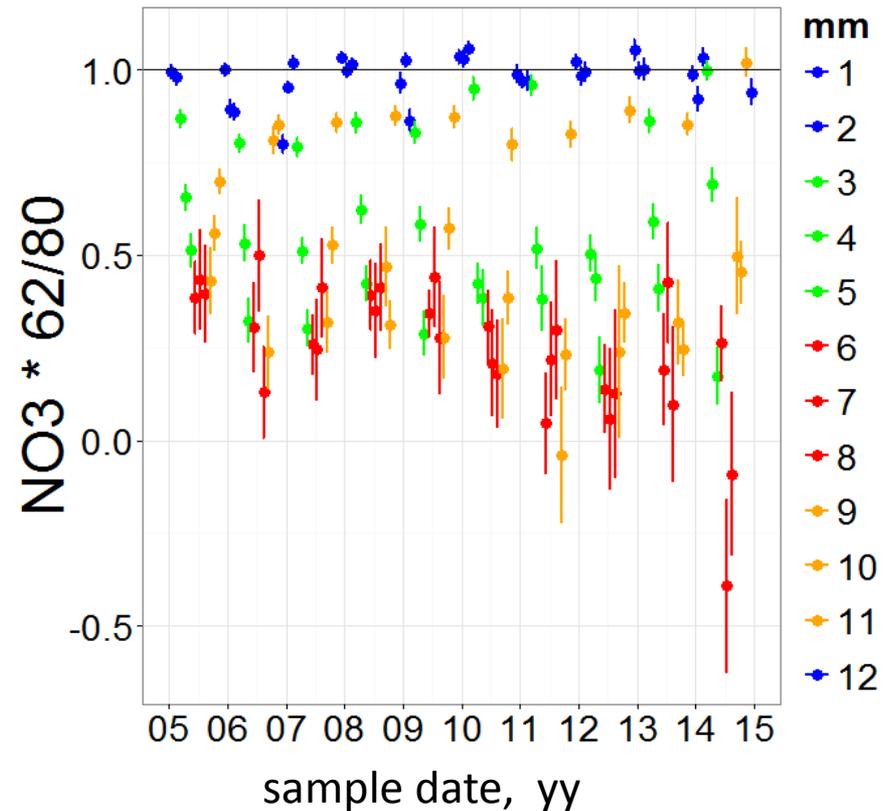
OLS coefficients from monthly regressions, m^2/g or g/g

The coefficients for nitrate are well determined for the winter months and behave as expected for ammonium nitrate. In the summer months, ammonium nitrate is lost from the PTFE filters.

Fabs \sim EC+OC+Fe+CHL+NO₃+SO₄



MF-EC \sim OC+Fe+CHL+NO₃+SO₄

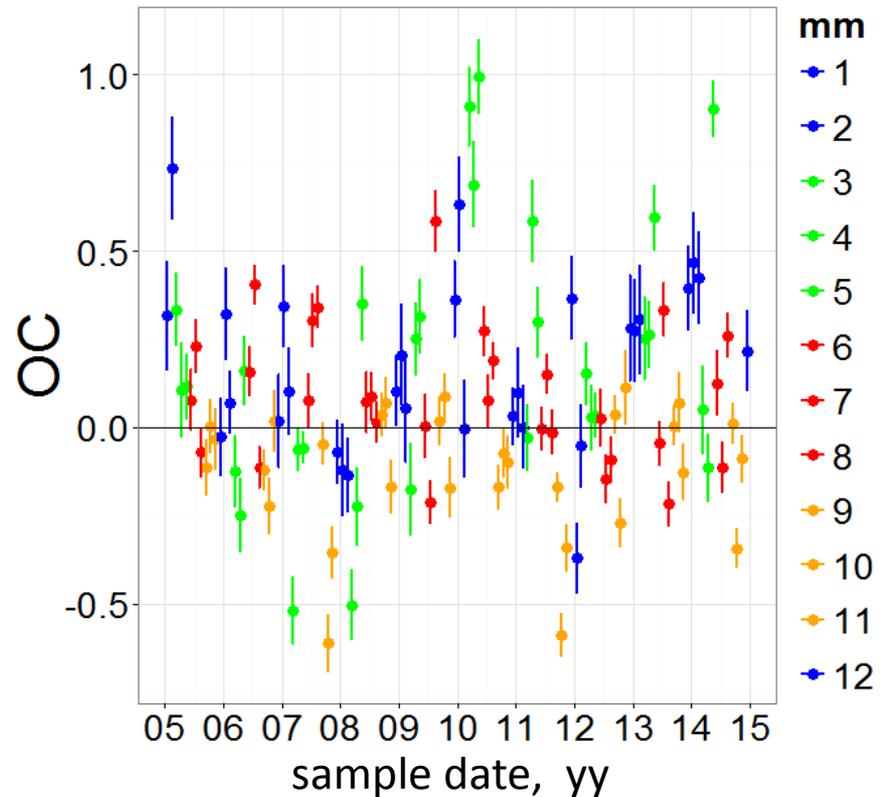
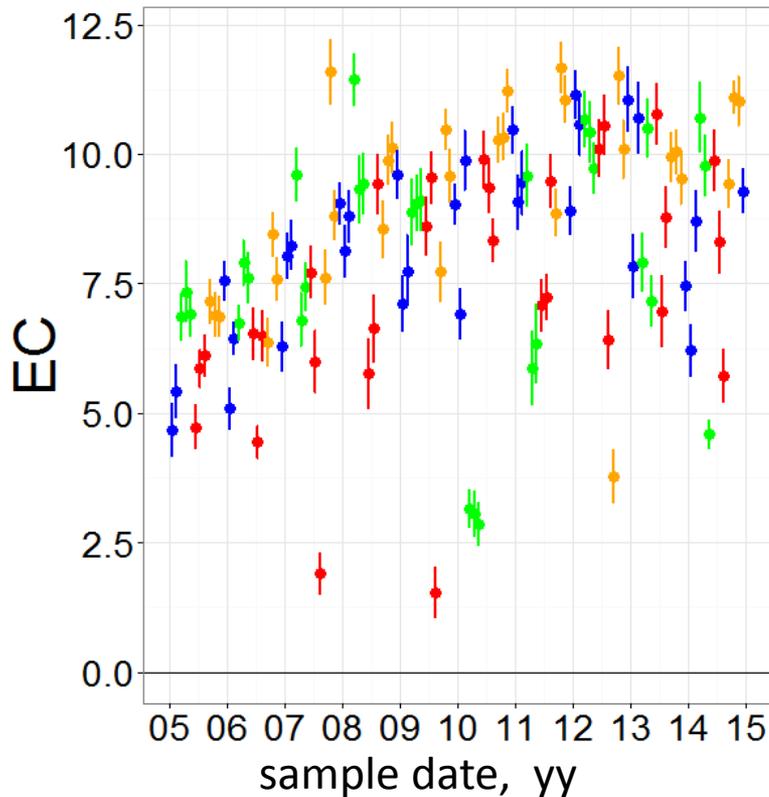


OLS coefficients from monthly regressions, m²/g or g/g

Regression does not yield stable mass coefficients for EC. The absorption coefficients for EC are accordingly paired below with those for OC, instead. These have not increased noticeably.

Fabs \sim EC+OC+Fe+CHL+NO3+SO4

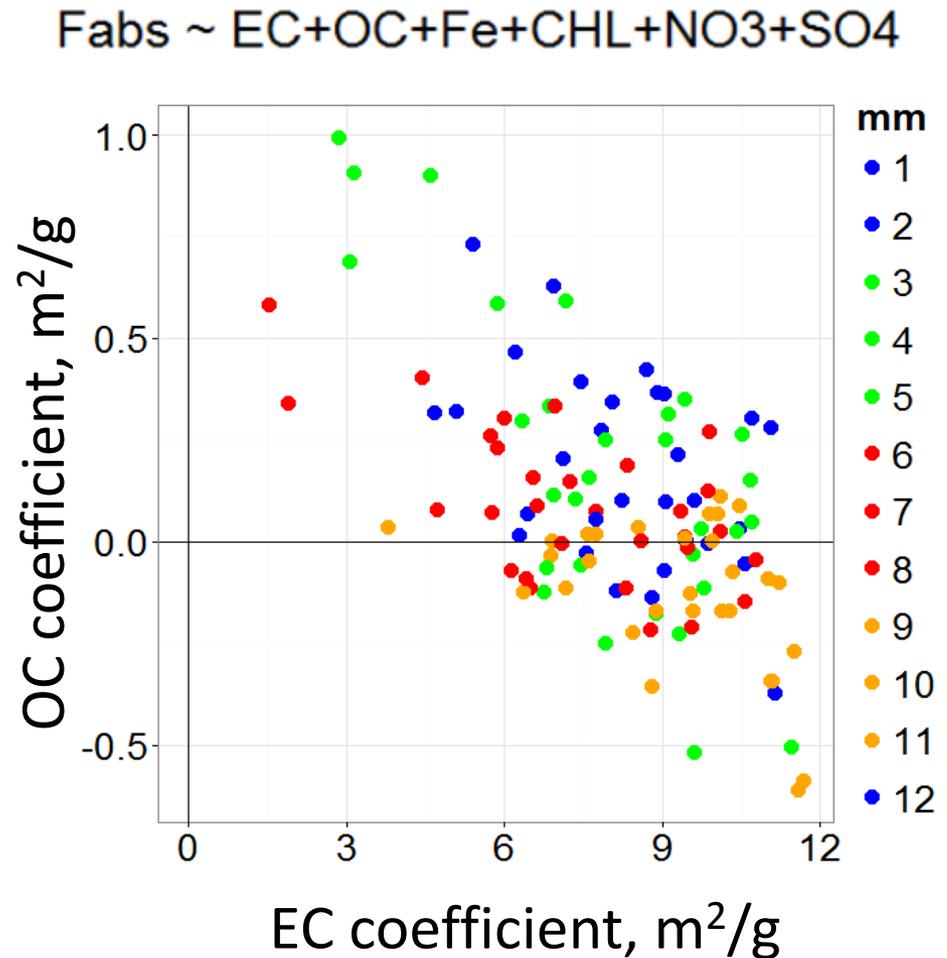
Fabs \sim EC+OC+Fe+CHL+NO3+SO4



OLS coefficients from monthly regressions, m^2/g

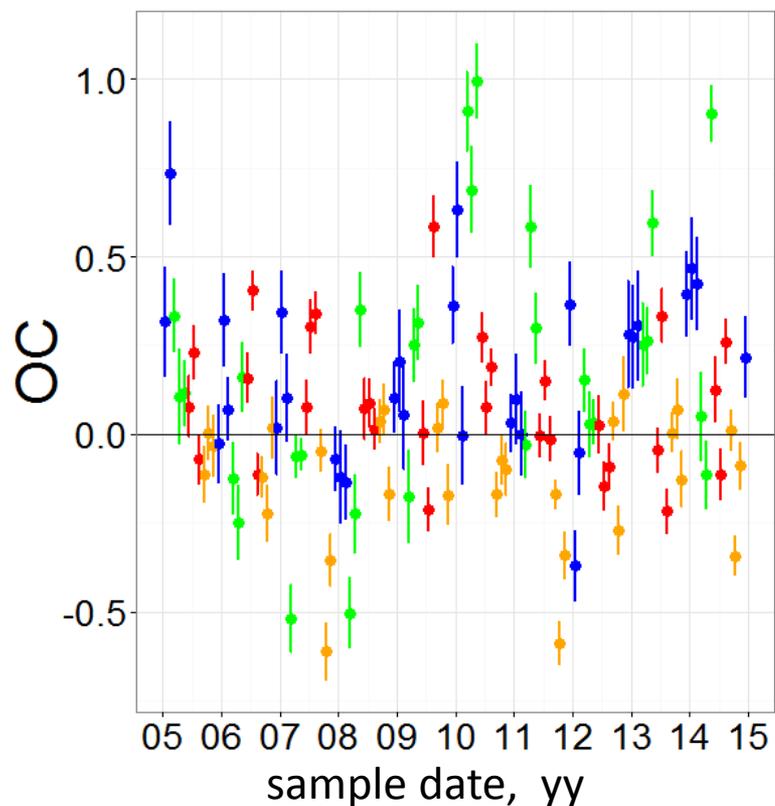
Interestingly, there is some anti-correlation (in all seasons) between the absorption coefficients attributed to EC and OC.

This coupling suggests uncertainty in the division of absorption between OC and EC, which could reflect statistical as well as measurement issues.

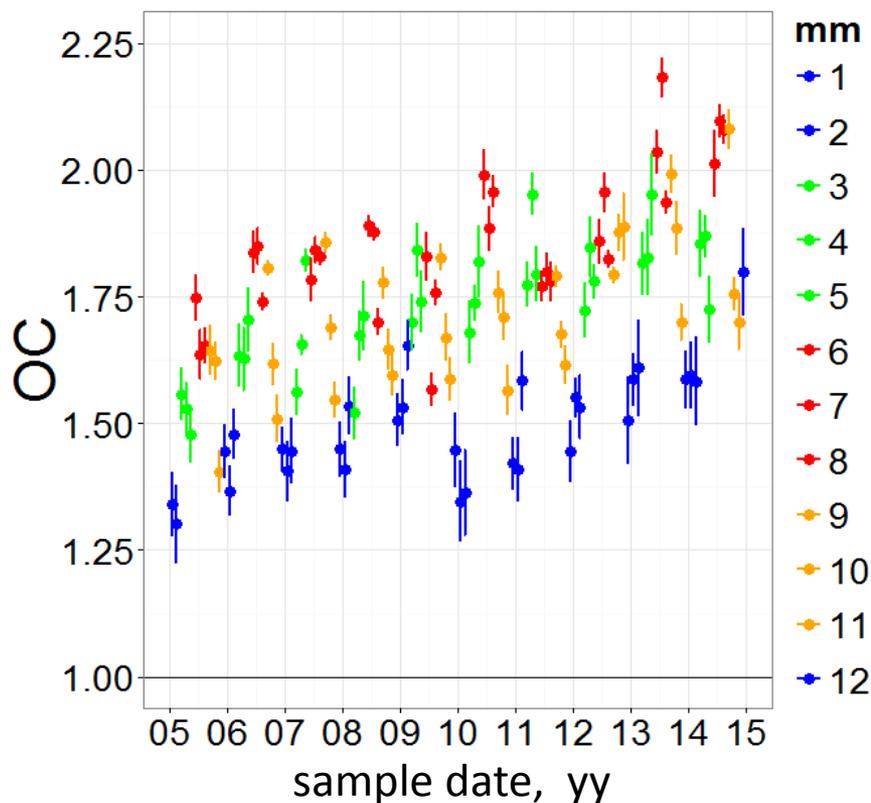


Mass-specific absorption by OC (m^2/g) is small, and may represent nothing more than statistical and analytical uncertainty in the OC-EC split. But note the clear trend in the mass ratio OM/OC (g/g)!

Fabs \sim EC+OC+Fe+CHL+NO₃+SO₄



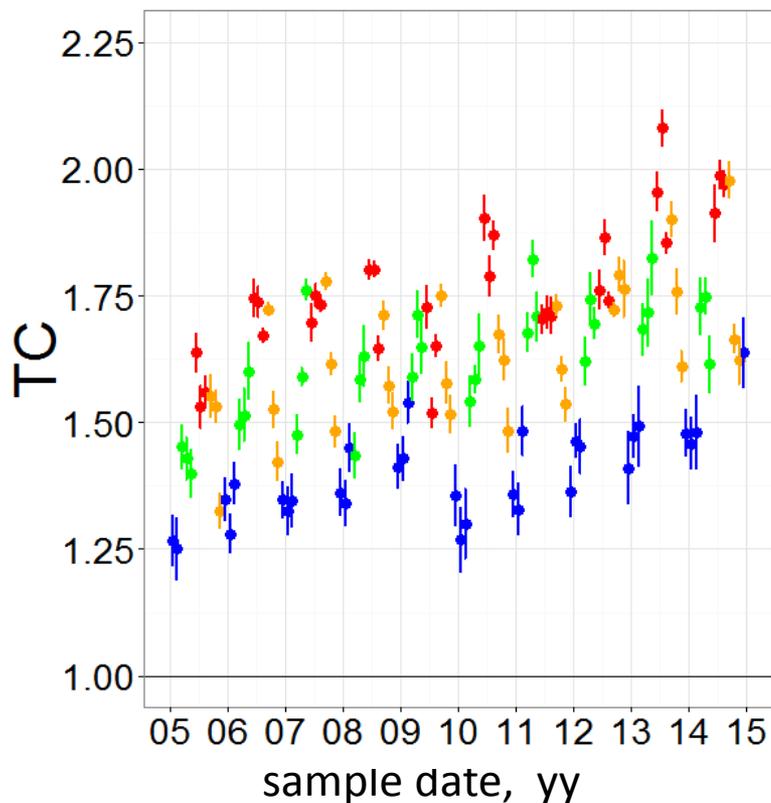
MF-EC \sim OC+Fe+CHL+NO₃+SO₄



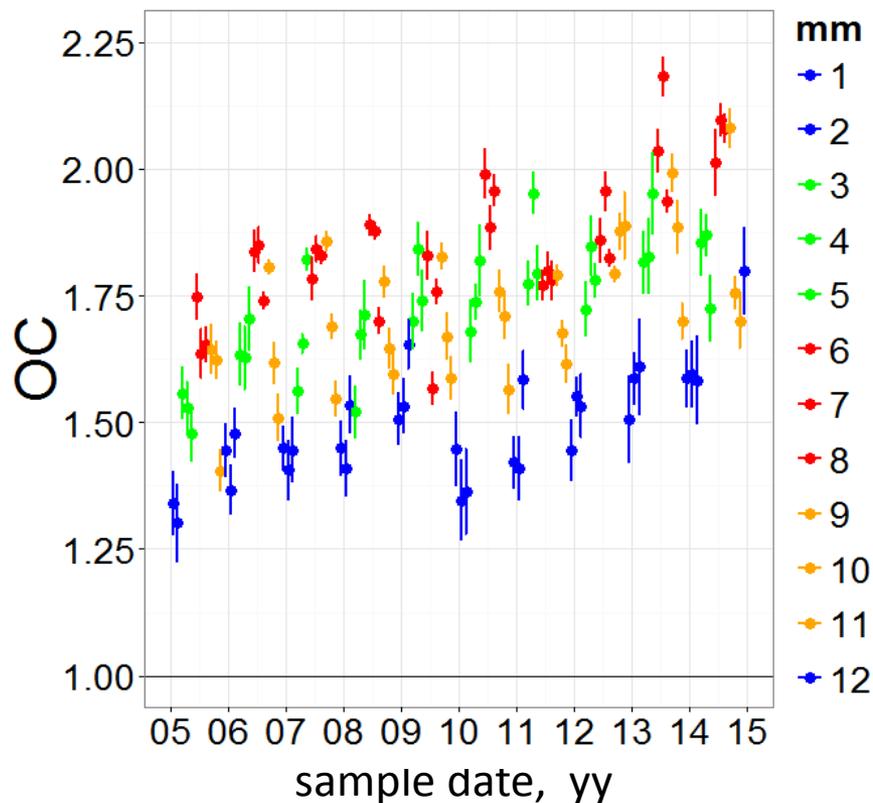
OLS coefficients from monthly regressions, m^2/g or g/g

The trend in OM/OC has nothing to do with the OC-EC split. Is the composition of the carbonaceous haze fraction evolving? Are there implications for the scattering attributed to OC in RHR tracking?

$$\text{MF} \sim \text{TC} + \text{Fe} + \text{CHL} + \text{NO}_3 + \text{SO}_4$$



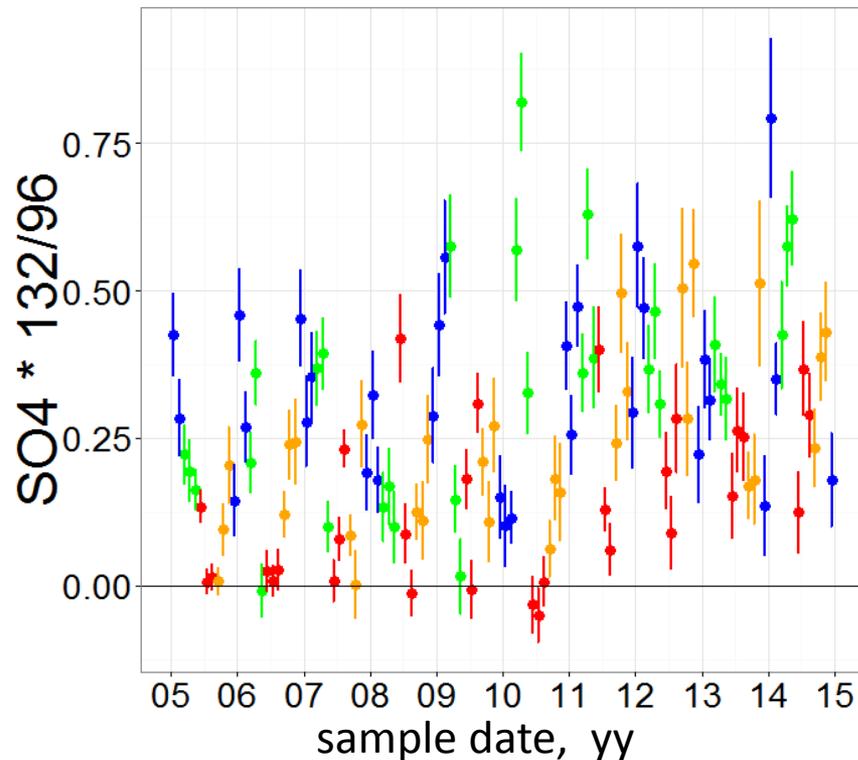
$$\text{MF-EC} \sim \text{OC} + \text{Fe} + \text{CHL} + \text{NO}_3 + \text{SO}_4$$



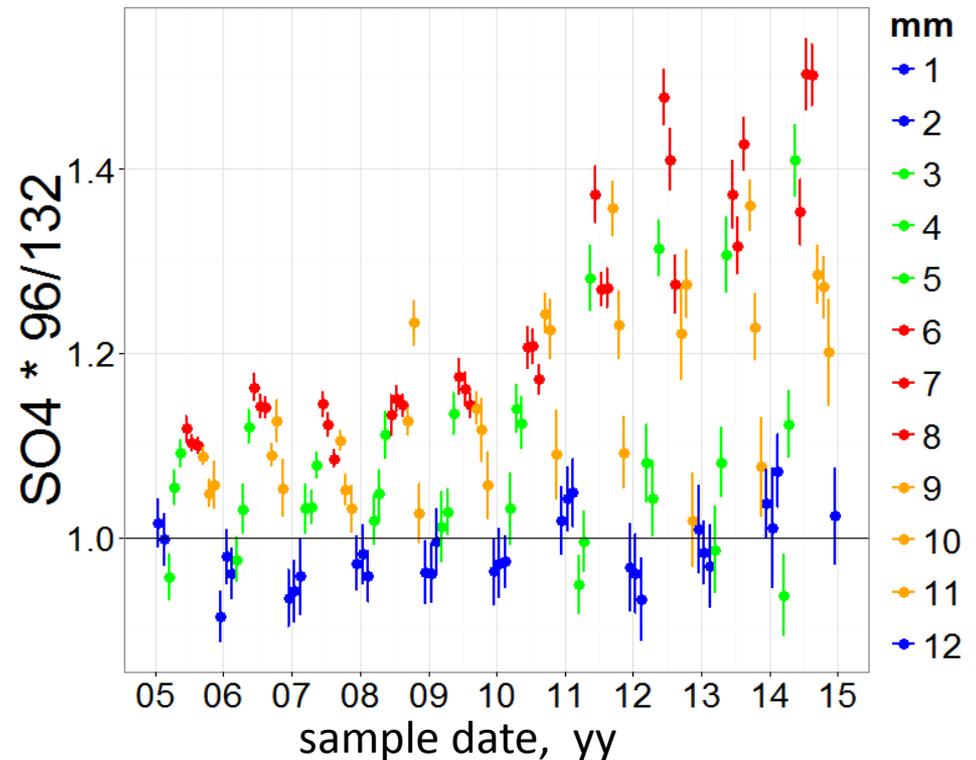
OLS coefficients from monthly regressions, g/g

A potential advisory/*caveat*: summer and late spring regression coefficients for hygroscopic salts increased starting in 2011, when sample handling and weighing were moved to a new lab space.

Fabs ~ EC+OC+Fe+CHL+NO₃+SO₄



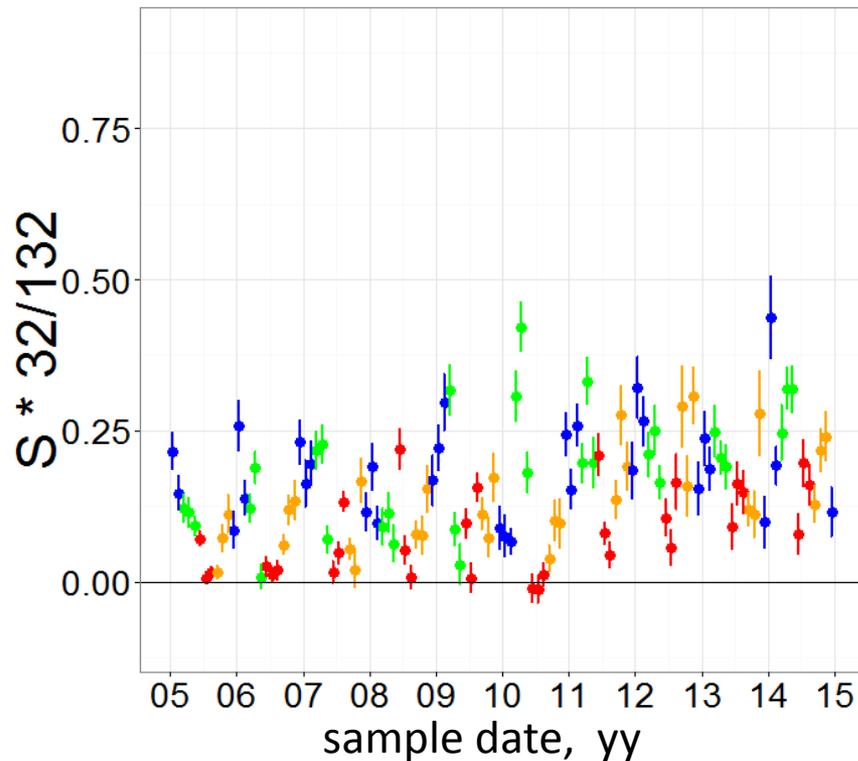
MF-EC ~ OC+Fe+CHL+NO₃+SO₄



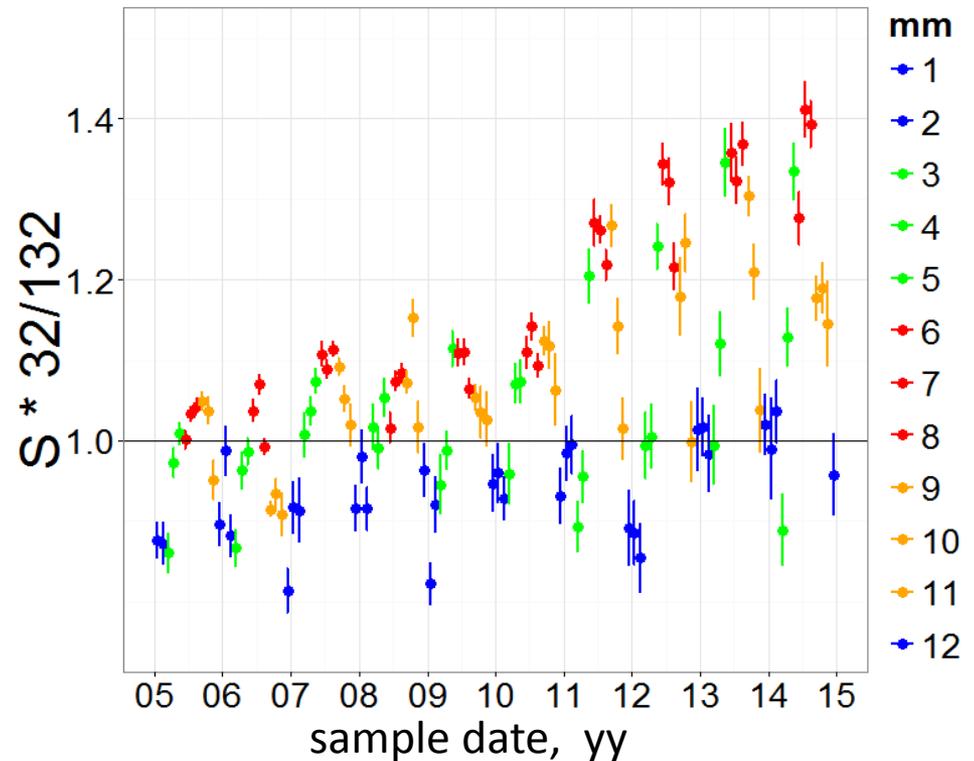
OLS coefficients from monthly regressions, m²/g or g/g

Data for XRF sulfur (below) and chloride ion (not shown) show the same behavior. A review of the new lab's environmental logs is in its early stages.

Fabs ~ EC+OC+Fe+CHL+NO3+S



MF-EC ~ OC+Fe+CHL+NO3+S



OLS coefficients from monthly regressions, m^2/g or g/g

$$MF - EC \sim OC + soil + CHL + NO3 + SO4$$

